



Standard Test Method for Saponification Number (Empirical) of Synthetic and Natural Waxes¹

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1. Scope

1.1 This test method covers the determination of the saponification number of synthetic waxes and natural waxes.

1.2 This test method is applicable to Fischer Tropsche, microcrystalline, polyethylene, and Montan Ester waxes.

1.3 Certain synthetic waxes, notably copolymers of ethylene, exhibit poor reproducibility when running saponification values. Reproducibility can be improved if cooking time in 7.2 is extended from 3 h to 18–20 h.

1.4 Some oxidized polyethylene and other waxes with a melt temperature above 100°C may give poor reproducibility.

1.5 Some dark-colored (Gardner Color 14) waxes may obscure the color change of the indicator, resulting in poor reproducibility.

1.6 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.7 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

E200 Practice for Preparation, Standardization, and Storage of Standard and Reagent Solutions for Chemical Analysis

3. Terminology

3.1 *Definitions:*

¹ This test method is under the jurisdiction of ASTM Committee D21 on Polishes and is the direct responsibility of Subcommittee D21.02 on Raw Materials.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

3.1.1 *saponification number*—the number of milligrams of potassium hydroxide required to hydrolyze 1 g of the sample and is a measure of the amount of saponifiable matter present.

4. Significance and Use

4.1 This test method is used to determine the property of ester functionality. Ester functionality determines the utility of the wax as well as being a significant quality control test.

5. Apparatus

5.1 *Analytical Balance.*

5.2 *Boiling Chips*, chemically resistant glass.

5.3 *Burets*, two 50-mL capacity with 0.1-mL graduations.

5.4 *Erlenmeyer Flasks*, 250-mL, alkali-resistant.

5.5 *Hot Plate.*

5.6 *Reflux Condenser.*

6. Reagents and Materials

6.1 *Purity of Reagents*—Reagent-grade chemicals or equivalent as specified in Practice E200 shall be used in all tests.

6.2 *Hydrochloric Acid Standard (0.5 N).*

6.3 *Phenolphthalein Indicator Solution (10 g/litre)*—Dissolve 1 g of phenolphthalein in 100 mL of USSD3A denatured ethanol or 95 % ethanol.

6.4 *Potassium Hydroxide, Alcoholic Solution (6.6 g/litre)*—Dissolve 6.6 g of potassium hydroxide (KOH) in USSD3A denatured ethanol or 95 % ethanol. Dilute to 1 L with the ethanol.

6.5 *Xylene.*

7. Procedure

7.1 Transfer approximately 1 g of the sample, weighed to the nearest 0.001 g to a 250-mL Erlenmeyer flask.

7.2 Add 40 mL of xylene and a few boiling chips to the flask. Dissolve by heating on the hot plate to the boiling point of xylene. As soon as the wax dissolves, remove from the hot